

Article

An ice core perspective on mercury pollution during the past 600 years

Samuel Beal, Erich C. Osterberg, Chris Zdanowicz, and David Fisher

Environ. Sci. Technol., Just Accepted Manuscript • DOI: 10.1021/acs.est.5b01033 • Publication Date (Web): 26 May 2015 Downloaded from http://pubs.acs.org on May 31, 2015

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



Environmental Science & Technology is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.



Location of Mount Logan (red triangle) in the St. Elias Mountains on the border of Alaska and Yukon, Canada. Arrows represent annual average vector wind at 500 mb from AD 1948–1998.⁴² 69x50mm (300 x 300 DPI)



20th Century trends in atmospheric Hg deposition at Mount Logan compared with ice core records of Pb pollution and the longest available measurements of remote atmospheric Hg concentrations. a) Mount Logan HgT fluxes (blue points) with 1σ error bars and LOESS smoother (red line). Grey shading denotes two periods of elevated HgT fluxes. b) Annually averaged Pb concentrations in the Mount Logan ice core¹⁸ and the Eclipse Icefield ice core²⁸, both in the St. Elias Mountains, and in the Act2 ice core in southern Greenland²⁹. c) Modeled firn-air Hg0 concentrations in a core from Summit in central Greenland modified from Faïn et al.³⁰ d) Total filterable (particulate) Hg concentrations in air samples from Resolute in Arctic Canada modified from Li et al.³¹ 68x111mm (300 x 300 DPI)



Multi-century HgT records from ice cores compared with estimates of primary anthropogenic emissions used in recent global Hg models. a) Mount Logan HgT fluxes (blue points) with 1σ error bars, LOESS smoother (red line), and inset with adjusted y-axis for the Preindustrial Period. b) HgT fluxes in the Upper Fremont Glacier ice core calculated using an assumed constant accumulation rate of 800 kg m-2 a-1 modified from Schuster et al.¹⁴ c) Estimated primary anthropogenic Hg emissions from industrial and mining sources modified from Streets et al.⁷ and additional emissions from commercial Hg use modified from Horowitz et al.⁸

136x103mm (300 x 300 DPI)

An ice core perspective on mercury pollution during the past 600 years

Environmental Science and Technology

Samuel A. Beal^{*,1}, Erich C. Osterberg¹, Christian M. Zdanowicz², and David A. Fisher³

¹ Department of Earth Sciences, Dartmouth College, Hanover, New Hampshire, USA.

² Department of Earth Sciences, Uppsala University, Uppsala, Sweden.

³ Department of Earth Sciences, University of Ottawa, Ottawa, Ontario, Canada.

* Corresponding author: SA Beal; U.S. Army Corps of Engineers, Engineer Research and Development Center, Cold Regions Research and Engineering Laboratory; 72 Lyme Road, Hanover, NH 03755, United States; phone: 603-646-4125; email: samuel.beal@usace.army.mil

Keywords: mercury record; global mercury cycle; Mount Logan; metal pollution



TOC/Abstract Art

1 Abstract (150-200 words)

2 Past emissions of the toxic metal mercury (Hg) persist in the global environment, yet 3 these emissions remain poorly constrained by existing data. Ice cores are high-resolution 4 archives of atmospheric deposition that may provide crucial insight into past atmospheric Hg 5 levels during recent and historical time. Here we present a record of total Hg (Hg_T) in an ice core 6 from the pristine summit plateau (5340 m asl) of Mount Logan, Yukon, Canada, representing 7 atmospheric deposition from AD 1410 to 1998. The Colonial Period (~1603-1850) and North 8 American "Gold Rush" (1850–1900) represent minor fractions (8% and 14%, respectively) of 9 total anthropogenic Hg deposition in the record, with the majority (78%) occurring during the 10 20th Century. A period of maximum Hg_T fluxes from 1940 to 1975 coincides with estimates of 11 enhanced anthropogenic Hg emissions from commercial sources, as well as with industrial 12 emissions of other toxic metals. Rapid declines in Hg_T fluxes following peaks during the "Gold 13 Rush" and the mid-20th Century indicate that atmospheric Hg deposition responds quickly to 14 reductions in emissions. Increasing Hg_T fluxes from 1993 until the youngest samples in 1998 15 may reflect the resurgence of Hg emissions from unregulated coal burning and small-scale gold 16 mining.

17 1. Introduction

Atmospheric deposition of the globally distributed and toxic metal mercury (Hg) contaminates ecosystems throughout the world.¹ Hg is introduced to the environment by *primary* emissions from anthropogenic sources (e.g., coal burning, Hg mining, and silver and gold mining) and natural sources (e.g., volcanism and weathering). Primary emissions occur predominantly as gaseous Hg⁰, with lesser quantities of reactive Hg²⁺ species that are either

23	gaseous or bound to particulates (Hg_p) . ² With a residence time in the atmosphere of ~1 year, Hg^0
24	can be transported globally before oxidation to Hg^{2+} causes deposition to the land/ocean surface
25	in precipitation (wet deposition) or by particle settling (dry deposition). ³ Deposited Hg can be
26	reemitted to the atmosphere through secondary emissions (e.g., ocean-air exchange and biomass
27	burning), which actively cycles Hg between the ocean, land, and atmosphere. ^{4,5}
28	Recent works provide disparate accounts of past primary anthropogenic Hg emissions,
29	which are thought to have occurred since at least AD 1570. ⁶ Framing this disparity are emissions
30	estimates, based on historical records of metal production and industrial/commercial activity, and
31	sediment records from remote lakes with small undisturbed watersheds. Emissions estimates
32	indicate that mining during the Preindustrial Period (pre-AD 1850) and the North American
33	"Gold Rush" (AD 1850–1920) represent \sim 30% and \sim 27%, respectively, of total primary
34	anthropogenic Hg emissions, and that emission rates peaked in the 1970s. ^{7,8} Conversely,
35	sediment records from remote lakes generally support background rates of atmospheric Hg
36	deposition (~1 μ g m ⁻² a ⁻¹) until ~AD 1850 and a subsequent continuous increase to the present-
37	day.9 Uncertainties in past emissions propagate to global model determinations of anthropogenic
38	Hg in the modern environment and predictions of the environmental response to future emissions
39	scenarios.

40 Ice cores are archives of atmospheric Hg deposition over seasonal to millennial 41 timescales, preserving a purely atmospheric signal with high temporal resolution and excellent chronological control through annual layer counting and distinct event stratigraphy. Although an 42 unknown fraction of deposited Hg is reemitted to the atmosphere by photoreduction within the 43 upper 3 to 60 cm of snowpack,¹⁰ the remaining Hg that is preserved in glacial ice may reveal 44 45 important information on past changes in atmospheric Hg deposition. Records of total Hg (Hg_T)

Environmental Science & Technology

46 from ice cores have been developed since the 1970s, but work in the 1990s by Boutron et al. 47 using clean methods for sample handling in the field and laboratory demonstrated that many of these earlier records were contaminated by orders of magnitude.¹¹ Clean sampling of remote 48 glacial snow and ice typically reveals Hg_T concentrations from <0.02 to ~4 pg g⁻¹.¹¹⁻¹³ To date, 49 the only published ice core Hg_T record covering the entire industrial period is from Upper 50 Fremont Glacier in Wyoming, United States.¹⁴ This record has been used extensively in literature 51 on the global Hg cycle,¹⁵ and it contains several unique characteristics including a prominent 52 53 "Gold Rush" peak, short-lived (2 to 3-year) peaks associated with explosive volcanic events in 54 both the Northern and Southern Hemisphere, and an abrupt decline in Hg_T concentrations in the 1980s.¹⁴ However, the unique characteristics of the record, relatively high Hg_T concentrations 55 (1.2–35 pg g⁻¹), and the glacier's proximity to upwind historical mining and industrial Hg 56 57 sources have led to uncertainty about the record's representativeness of global trends in atmospheric Hg.^{9,15} New ice core Hg_T records from remote glaciers could provide useful 58 59 additional constraints on past global Hg emissions. 60 Here we present a record of Hg_T from AD 1410 to 1998 in an ice core from Mount 61 Logan, Yukon, Canada. The Mount Logan ice core, collected from the mountain's summit 62 plateau (5340 m asl), preserves deposition from the free troposphere sourced by winds traversing

the North Pacific Ocean (Figure 1).^{16,17} The core's location, tightly constrained chronology, and
 high-resolution co-registered geochemistry¹⁸ allow us to interpret the Mount Logan Hg_T record
 and assess its relevance to past global Hg emissions.

2. Materials and Methods 66

2.1. Core Collection 67

68 The Mount Logan ice core was collected from Prospector-Russell Col on the summit 69 plateau (5340 m asl) of Mount Logan in the summers of 2001 and 2002 by the Geological 70 Survey of Canada. The coring site is near the center of dividing glacier flow with low ice velocities ($<0.2 \text{ m a}^{-1}$), and the site has persistent below-freezing temperatures based on 71 automatic weather station data (mean annual: -29°C) and the lack of refrozen surface melt layers 72 in the core.^{19,20} The core was collected from a dry borehole using an electromechanical drill from 73 74 2 m depth until bedrock at 186 m depth. We focus on the depth range of 3 to 128 m that 75 represents net snow accumulation between AD 1410 and 1998. Previous work established the depth-age model for this ice core based on annual layer counting of seasonal oscillations in δ^{18} O 76 77 and Na and U concentrations between 1700 and 1998 (3-109 m depth) and an ice flow model constrained by well-dated volcanic signals between 1410 and 1700 (109–128 m depth).^{18,21,22} 78

79

2.2. Core Decontamination

80 We sampled discrete sections of the ice core, with each section comprising at least one complete year of accumulation. Within the upper 100 m of the ice core, which included firn (3-81 82 45 m depth) and ice (45–100 m depth), the volume of material for each section required division 83 into multiple samples that we plot individually. We melted firn and ice samples in 500-ml glass jars with PTFE septa and polypropylene caps which we pre-cleaned following Method B in 84 Hammerschmidt et al.²³ Briefly, this cleaning method involved successive storage in a Citranox 85 86 and >18.2 M Ω deionized (DI) water solution for 6 days, in 10% Trace-Metal Grade HCl for 6 87 days, and in 1% BrCl (made from reagent grade KBr, KBrO₃, and Optima HCl) for 1 day. All 88 sample work was conducted in a HEPA-filtered laminar flow bench with a double layer of

89	activated carbon cloth covering the filter air intake. We cut firn and ice pieces that were longer
90	than 10 cm with a stainless steel saw to accommodate the height of the glass jars. For firn
91	samples (3–45 m), we manually removed the outer surfaces of each sample using ZrO ceramic
92	knives (pre-cleaned with 10% HCl) in a -10 °C freezer before transferring the decontaminated
93	firn to the pre-cleaned glass jars. For ice samples (45–128 m), we removed potentially
94	contaminated outer ice surfaces by partially melting each sample for 1.5 hours at room
95	temperature in Whirlpak bags before rinsing with DI water and transferring to the pre-cleaned
96	glass jars. We tested contamination of outer ice core surfaces by partially melting selected ice
97	samples in pre-cleaned glass jars instead of Whirlpak bags, and then treating the cleanly
98	collected meltwater from the outer core surfaces as samples. We made procedural blanks by
99	freezing low-Hg ($<0.02 \text{ pg g}^{-1}$) DI water in pre-cleaned glass jars and processed them following
00	the decontamination methods for either firn or ice. Before melting, we added concentrated BrCl
01	to each sample or procedural blank to a final concentration of 0.1% BrCl. ²⁴

102 **2.3. Hg**_T Analysis

103 Within 72 hours of melting, we measured Hg_T in ~24 ml sample aliquots using Purge and 104 Trap with Cold Vapor Atomic Fluorescence Spectrometry (Brooks Rand Automated Total Mercury Analyzer) following EPA Method 1631E. Hg_T refers to all BrCl-oxidizable forms of Hg 105 present, including Hg^{2+} , Hg^{0} , organic Hg, and Hg adsorbed to particulates.²⁴ The instrumental 106 detection limit was 0.02 pg g⁻¹, but the method detection limit (MDL; determined here as 3σ of 107 the procedural blanks) ranged from 0.01 to 0.24 pg g^{-1} (median 0.06) depending on the method 108 109 and day of sampling. Sample concentrations were considered detectable if their procedural blank 110 subtracted concentrations exceeded the MDL.

111 2.4. Flux Calculations

112 We determine Hg_T fluxes as the product of measured Hg_T concentrations and water 113 equivalent accumulation rates (ARs). In the Mount Logan ice core, ARs are calculated within the sub-annually resolved portion of the core (1700–1998) from measurements of density (kg m⁻³) 114 and annual layer thickness (m a⁻¹), and ARs are estimated within the annually resolved portion of 115 116 the core (1410–1700) as the average of measured ARs from the sub-annually resolved portion $(420 \pm 90 \text{ kg m}^{-2} \text{ a}^{-1})$ because annual layer thickness is not accurately determined at these 117 depths.¹⁸ 118

119

3. Results and Discussion

120 Of the 130 samples that we melted, 88 samples had detectable Hg_T concentrations (Table 121 1 and Figure S1). Table 1 shows the occasionally high and variable procedural blank values that 122 limited detection of Hg_T in the remaining samples. We suspect that day-to-day changes in air 123 quality within the HEPA-filtered laminar flow benches are responsible for this variability, 124 highlighting the need for the many procedural blanks used in this study. Among the samples with 125 detectable Hg_T concentrations, preservation of a signal from atmospheric Hg deposition is supported by similar Hg_T concentrations found in snow/ice from other remote glaciers^{11–13} and 126 127 by the order of magnitude concentration difference between the samples and the core exterior 128 removed during decontamination (Table 1).

In the most recent samples from 1994 to 1998, Hg_T fluxes (0.19–0.52 µg m⁻² a⁻¹) are an 129 130 order of magnitude lower than recent measurements of wet deposition across the western United States (~1–9 μ g m⁻² a⁻¹) and in sub-Arctic Alaska and Canada (~1–5 μ g m⁻² a⁻¹).^{25,26} This 131 difference could be attributed to the relative inefficiency of gaseous Hg²⁺ scavenging by snow,²⁷ 132 133 which is the only phase of precipitation at Mount Logan, and to the reemission of a fraction of

134 deposited Hg from the snowpack. Repeated sampling of surface snows generally indicate that 135 >50% of Hg is reemitted within 24 hours, although there is great variability both within individual study sites and between geographic areas.¹⁰ Nevertheless, we propose that multi-136 137 annual variations in the Mount Logan Hg_T flux record reflect past changes in mid-tropospheric atmospheric Hg^{2+} deposition based on several lines of evidence. First, the estimated Hg^{0} content 138 139 in firn/ice air can account for only 0.2 to 7% of the measured Hg_T concentrations (see Supporting 140 Information). Hg_T concentrations are only weakly correlated with co-registered lithogenic 141 element and Pb concentrations (r = 0.23-0.35, p < 0.05), which are proxies for Hg_p from crustal 142 and industrial sources, respectively (Table S1). Hg_T concentrations are neither correlated with 143 co-registered chloride concentrations, nor with ARs (Table S1), both of which are thought to affect Hg reemission from the snowpack.^{10,28} Finally, the Mount Logan Hg_T flux record shows 144 145 striking similarities with two of the longest published records of remote atmospheric Hg levels (Figure 2) and with historical changes in Hg use, which we discuss below. On shorter (sub-146 147 annual to annual) timescales, the relatively high variability, or spikiness, of the Mount Logan 148 Hg_T flux record is likely the result of factors influencing Hg reemission from the snowpack. 149 Accordingly, we use a LOESS smoothed line to filter out this short-term variability and highlight 150 longer-term variations in the Mount Logan Hg_T flux record.

151 **3.1. The 20th Century**

Figure 2 shows the Mount Logan Hg_T record during the 20th Century, revealing nearbackground Hg_T fluxes ($0.02-0.06 \ \mu g \ m^{-2} \ a^{-1}$) from 1900 to 1940, a broad peak between ~1940 and 1975 (maximum of 1.2 $\ \mu g \ m^{-2} \ a^{-1}$ in 1960), decreasing fluxes during the 1980s, and a final rise through the 1990s. Comparison with high resolution Pb records from the Mount Logan ice core and other high northern latitude glaciers provide information on both the source and

157 speciation of Hg_T in the Mount Logan ice core. The mid-century peak in Mount Logan Hg_T 158 fluxes is coincident with a period of maximum Pb deposition registered in ice cores from 159 Greenland and from Eclipse Icefield (located ~50 km northwest of Mount Logan at 3017 m asl), 160 attributed to industrial (e.g., coal burning) aerosol emissions from North America and Europe 161 that increased from World War II until the enactment of environmental regulations in the 1970s (Figure 2b).^{29,30} However, North American and European Pb aerosols did not reach the Mount 162 163 Logan summit plateau during this time (Figure 2b) because of the relatively short (7-10 days) atmospheric residence time of Pb,³¹ and the persistent air mass source from the west (North 164 Pacific and Asia) due to the site's elevation in the free troposphere.¹⁸ Different emissions sources 165 166 for 20th Century Pb and Hg deposited on Mount Logan is also supported by the lack of 167 correlation between the two co-registered records (Table S1). Rather, the mid-century peak in Mount Logan Hg_T fluxes mirrors a record of atmospheric Hg⁰ concentrations from firn-air at 168 169 Summit Greenland (Figure 2c). Altogether, these comparisons suggest a coherent change in high 170 northern latitude atmospheric Hg levels during the 20th Century associated with industrial 171 emissions from North America and Europe.

172 Subsequent to the mid-century peak, Hg_T fluxes decline during the 1980s and approach near-background fluxes (0.05–0.13 μ g m⁻² a⁻¹) between 1989 and 1991. Decreases during this 173 time are similarly observed in the firn-air Hg⁰ record from Summit Greenland³² and in air 174 measurements of Hg_p at Resolute Bay in the Canadian High Arctic (Figure 2d).³³ A record-175 176 maximum peak in Mount Logan Pb concentrations between 1980 and 1989, likely resulting from the transport and deposition of industrial aerosols from Asia,¹⁸ is coincident with this period of 177 depressed Mount Logan Hg_T fluxes and therefore indicative of negligible Hg_p contributions from 178 179 Asia during this time. We sampled the Mount Logan ice core at continuous 0.5-year resolution

between late 1989 and late 1991 to examine a possible Hg_T pulse from the April 2, 1991 eruption
of Mount Pinatubo that sent volcanic ash and sulfur-containing aerosols into the stratosphere.³⁴
Such a pulse is not evident in either the Mount Logan or the Upper Fremont Glacier records
during the timing of this eruption.¹⁴ A number of other major volcanic eruptions, including ones
in the Southern Hemisphere, are identified as significant Hg_T spikes, in the Upper Fremont
Glacier record,¹⁴ but material from the Mount Logan ice core covering these events has been
entirely consumed for other studies.

Increasing Mount Logan Hg_T fluxes from 1993 until 1998 (the age of the youngest samples) coincide with estimated increasing emissions of Hg^0 , Hg_p , and Pb from coal combustion in Asia.^{7,35} The trans-Pacific transport and deposition of industrial aerosols from Asia are thought to be responsible for increasing Pb concentrations in the Mount Logan ice core during the 1990s (Figure 2b).¹⁸ We therefore infer that the increase in Hg_T fluxes between 1993 and 1998 is due, at least in part, to deposition of Hg_p associated with Asian industrial emissions.

193

3.2. The "Gold Rush" (AD 1850–1900)

194 Figure 3 displays the entire 600-year Mount Logan record, which shows, in addition to 195 the mid-20th Century peak, a smaller irregular peak in Hg_T fluxes that begins in the 1860s, reaches a maximum of 0.32 μ g m⁻² a⁻¹ in ~1883, and remains elevated until a return to 196 197 background at the turn of the century. This 19th Century peak is largely synchronous with an 198 estimated peak in primary Hg emissions from silver and gold mining in North America, and 199 associated Hg production (mainly in Spain, Slovenia, and Italy), during the "Gold Rush" between 1860 and 1920 (Figure 3c).⁷ The Upper Fremont Glacier Hg_T record (Figure 3b) also 200 201 shows a "Gold Rush" peak, which the authors suggest is caused by the regional transport and deposition of Hg from gold mining in upwind California.¹⁴ The synchronous response of the 202

203 Mount Logan Hg_T flux record to estimated global Hg emissions during the "Gold Rush", and the 204 lack of major upwind Hg point sources during the time, support a distinct anthropogenic change 205 in high northern latitude Hg levels caused by the "Gold Rush". However, the Mount Logan and 206 Upper Fremont Glacier "Gold Rush" Hg_T peaks are both substantially smaller than their 207 respective 20th Century peaks, whereas emissions estimates indicate that the "Gold Rush" 208 included an extended period of some of the highest primary anthropogenic Hg emissions of the past 400 years (Figure 3).^{7,8} Lower-than-estimated anthropogenic Hg emissions during the "Gold 209 Rush" may explain the lack of such a signal in many lake sediment records.⁹ 210

211

3.3. The Preindustrial Period (AD 1410–1850)

The inset of Figure 3a shows stable background Mount Logan Hg_T fluxes (0.015–0.023 212 μ g m⁻² a⁻¹) from ~1410 to ~1561, followed by a period of increased low-level variability in Hg_T 213 fluxes lasting until ~1850, when the impact of industrial emissions becomes clearly apparent. 214 The introduction in 1570 of liquid Hg⁰ for silver mining in Colonial Spanish South America 215 likely marked the first global distribution of anthropogenic Hg,^{6,36,37} so we interpret the Mount 216 217 Logan Hg_T fluxes prior to 1570 as representative of natural deposition and subsequent variations 218 from this background as due to anthropogenic Hg emissions. The background pre-Colonial Hg_T fluxes are in good agreement with a few measurements of Hg_T (0.009 μ g m⁻² a⁻¹) during the early 219 and middle Holocene in an ice core from Dome C, Antarctica.³⁸ The low-level increases in 220 221 Mount Logan Hg_T fluxes between ~1600 and 1850 suggest that Hg emissions from Colonial 222 silver mining were globally transported, however Hg_T fluxes during this time are at least an order 223 of magnitude lower than Hg_T fluxes during both the "Gold Rush" and the late 20th Century.

Page 15 of 23

224 **3.4. Potential Global Hg Cycle Implications**

225 Integrating the smoothed Mount Logan Hg_T flux record (Figure 3a) after subtracting 226 average background (1410–1561) fluxes yields ~78% of anthropogenic Hg deposition registered 227 by the ice core occurring during the 20th Century, ~14% during the "Gold Rush" (1850–1900), 228 and ~8% during the Colonial Period (~1603–1850). This 20th Century dominance in the Mount 229 Logan record differs significantly from estimates of primary anthropogenic Hg emissions from 230 industrial, mining, and commercial sources, which combined show that the "Gold Rush" and 231 Preindustrial Period contributed ~27% and ~30% of total anthropogenic emissions, respectively.^{7,8} Hg releases to water and soil may emit additional anthropogenic Hg to the 232 233 atmosphere, but estimates of these releases generally scale with total (commercial, industrial, and mining) primary emissions since 1850.⁸ The potential for secondary emissions from the oceans 234 235 and land to influence deposition across the time periods discussed here make this a first-order 236 comparison. However, the relatively small anthropogenic Hg deposition registered by the Mount 237 Logan ice core before 1900 suggests that current estimates of Hg emissions from Colonial and 238 "Gold Rush" mining are erroneously high. In turn, global Hg models forced with these high 239 emissions likely overestimate the influence of early mining on anthropogenic Hg in the modern 240 global environment.

Rapid declines in the Hg_T flux records from both Mount Logan and Upper Fremont
Glacier coincide with estimated decreases in primary anthropogenic emissions from mining
following the "Gold Rush"⁷ and from commercial sources in the 1970s⁸ (Figure 3). However,
global Hg models forced with these emissions estimates show relatively minor decreases in
atmospheric Hg following the "Gold Rush" and the 1970s, which are the result of high emission
forcings from early mining and secondary emissions that are estimated as approximately

247	equivalent to primary emissions during the present-day. ^{$4,5,8,39$} Similar to the ice core Hg _T records,
248	the Summit Greenland firn-air Hg ⁰ record (Figure 2c) exhibits a rapid decrease following the
249	1970s, but the extent of its recovery is difficult to assess because of the record's brevity and the
250	potential for smoothing by diffusion of gaseous Hg ^{0.32} The rapid decreases in atmospheric Hg
251	deposition implied by the Mount Logan and Upper Fremont Glacier records may not be easily
252	detectable in lake sediment records due to lagged declines in watershed inputs. ⁴⁰ However,
253	decreases following both the "Gold Rush" and the 1970s are apparent in Hg flux records from
254	two remote lakes with exceptional characteristics: the crater lake Challa in eastern equatorial
255	Africa (900 m asl) that has almost no surficial watershed, ⁹ and the tarn Yanacocha in southern
256	Peru (4900 m asl) that has a small, poorly vegetated watershed composed predominantly of
257	exposed bedrock. ⁴¹ Further research is needed to constrain net exchanges between the land,
258	oceans, and atmosphere that cause anthropogenic Hg to persist in the environment.
259	Rising Hg _T fluxes during the most recent part of the Mount Logan record, from 1993 to
260	1998, may reflect a new period of increasing primary Hg emissions, driven by coal burning in
261	Asia and small-scale gold mining in developing countries around the world, that is estimated to
262	have continued at least through 2008. ^{2,7} Remarkably, however, the majority of atmospheric
263	measurements, primarily from the Atlantic near sea level, reveal declining Hg ⁰ concentrations
264	from ~2000 to 2009. ^{42,43} New high-resolution records of atmospheric Hg deposition, in addition
265	to air measurements, are needed to document trends in global atmospheric Hg pollution and to
266	address fundamental deficiencies in our current understanding of the global Hg cycle. The
267	collection of new ice cores for this work demands urgency as glaciers around the world are
268	currently experiencing accelerated mass losses, and many may soon vanish.44,45

269 Acknowledgments

- 270 This material is based upon work supported by the National Science Foundation under Grant No.
- 271 BCS-1232844. We thank the following collaborators: the Geological Survey of Canada for
- 272 making the Mount Logan ice core available; T. Overly and G. Wong for ice core processing in
- 273 Ottawa; Z. Courville for freezer access; B. Jackson and C. Lamborg for scientific insight; and J.
- 274 Creswell and V. Engel at Brooks Rand Labs for instrument assistance. We also thank four
- anonymous reviewers.

276 Supporting Information Available

- 277 The primary data for this paper can be accessed in the Supporting Information. This information
- is available free of charge via the Internet at http://pubs.acs.org.

279 **References**

- (1) Fitzgerald, W. F.; Engstrom, D. R.; Mason, R. P.; Nater, E. A. The Case for Atmospheric
 Mercury Contamination in Remote Areas. *Environ. Sci. Technol.* 1998, *32* (1), 1–7.
- 282 (2) Pacyna, E. G.; Pacyna, J. M. Global Emission of Mercury from Anthropogenic Sources in
 283 1995. *Water Air Soil Poll* 2002, *137* (1), 149–165.
- (3) Lamborg, C. H.; Fitzgerald, W. F.; Damman, A. W. H.; Benoit, J. M.; Balcom, P. H.;
 Engstrom, D. R. Modern and historic atmospheric mercury fluxes in both hemispheres:
 Global and regional mercury cycling implications. *Global Biogeochem.Cycles* 2002, *16*(4), 51–1 51–11.
- (4) Friedli, H. R.; Arellano, A. F.; Cinnirella, S.; Pirrone, N. Initial Estimates of Mercury
 Emissions to the Atmosphere from Global Biomass Burning. *Environ.Sci.Technol.* 2009,
 43 (10), 3507–3513.
- (5) Soerensen, A. L.; Sunderland, E. M.; Holmes, C. D.; Jacob, D. J.; Yantosca, R. M.; Skov,
 H.; Christensen, J. H.; Strode, S. A.; Mason, R. P. An Improved Global Model for Air-Sea
 Exchange of Mercury: High Concentrations over the North Atlantic. *Environ. Sci. Technol.* 2010, 44 (22), 8574–8580.
- 295 (6) Nriagu, J. O. Legacy of mercury pollution. *Nature* **1993**, *363* (6430), 589–589.
- (7) Streets, D. G.; Devane, M. K.; Lu, Z.; Bond, T. C.; Sunderland, E. M.; Jacob, D. J. All Time Releases of Mercury to the Atmosphere from Human Activities. *Environ Sci Technol* 2011, 45 (24), 10485–10491.
- (8) Horowitz, H. M.; Jacob, D. J.; Amos, H. M.; Streets, D. G.; Sunderland, E. M. Historical
 Mercury Releases from Commercial Products: Global Environmental Implications.
 Environ Sci. Technol. 2014, 48 (17), 10242, 10250
- 301 *Environ. Sci. Technol.* **2014**, *48* (17), 10242–10250.

302 (9) Engstrom, D. R.; Fitzgerald, W. F.; Cooke, C. A.; Lamborg, C. H.; Drevnick, P. E.; 303 Swain, E. B.; Balogh, S. J.; Balcom, P. H. Atmospheric Hg emissions from preindustrial 304 gold and silver extraction in the Americas: a reevaluation from lake-sediment archives. 305 Environ Sci Technol 2014, 48 (12), 6533-6543. (10) Durnford, D.; Dastoor, A. The behavior of mercury in the cryosphere: A review of what 306 307 we know from observations. J Geophys Res 2011, 116, D06305. 308 (11) Boutron, C. F.; Vandal, G. M.; Fitzgerald, W. F.; Ferrari, C. P. A forty year record of 309 Mercury in central Greenland snow. Geophys Res Lett 1998, 25 (17), 3315–3318. 310 Zdanowicz, C.; Krümmel, E. M.; Lean, D.; Poulain, A. J.; Yumvihoze, E.; Chen, J.; (12)311 Hintelmann, H. Accumulation, storage and release of atmospheric mercury in a glaciated 312 Arctic catchment, Baffin Island, Canada. Geochim Cosmochim Ac 2013, 107 (0), 316-313 335. 314 (13) Zheng, J.; Pelchat, P.; Vaive, J.; Bass, D.; Ke, F. Total mercury in snow and ice samples 315 from Canadian High Arctic ice caps and glaciers: A practical procedure and method for total Hg quantification at low pg g-1 level. Sci Total Environ 2014, 468-469 (0), 487-316 317 494. 318 (14)Schuster, P. F.; Krabbenhoft, D. P.; Naftz, D. L.; Cecil, L. D.; Olson, M. L.; Dewild, J. F.; 319 Susong, D. D.; Green, J. R.; Abbott, M. L. Atmospheric Mercury Deposition during the 320 Last 270 Years: A Glacial Ice Core Record of Natural and Anthropogenic Sources. Environ Sci Technol 2002, 36 (11), 2303–2310. 321 322 (15) UNEP. Global Mercury Assessment 2013: Sources, Emissions, Releases and 323 Environmental Transport; DTI/1636/GE; UNEP Chemicals Branch: Geneva, Switzerland, 324 2013. 325 (16) Zdanowicz, C.; Hall, G.; Vaive, J.; Amelin, Y.; Percival, J.; Girard, I.; Biscaye, P.; Bory, 326 A. Asian dustfall in the St. Elias Mountains, Yukon, Canada. Geochim Cosmochim Ac 327 **2006**, 70 (14), 3493–3507. (17) Zdanowicz, C.; Fisher, D.; Bourgeois, J.; Demuth, M.; Zheng, J.; Mayewski, P.; Kreutz, 328 329 K.; Osterberg, E.; Yalcin, K.; Wake, C.; et al. Ice Cores from the St. Elias Mountains, 330 Yukon, Canada: Their Significance for Climate, Atmospheric Composition and Volcanism 331 in the North Pacific Region. ARCTIC; Vol 67, No 5 (2014): Supplement 1: 1-107 2014. 332 (18) Osterberg, E.; Mayewski, P.; Kreutz, K.; Fisher, D.; Handley, M.; Sneed, S.; Zdanowicz, 333 C.; Zheng, J.; Demuth, M.; Waskiewicz, M.; et al. Ice core record of rising lead pollution 334 in the North Pacific atmosphere. Geophys Res Lett 2008, 35 (5), L05810. 335 (19) Fisher, D. A.; Wake, C.; Kreutz, K.; Yalcin, K.; Steig, E.; Mayewski, P.; Anderson, L.; 336 Zheng, J.; Rupper, S.; Zdanowicz, C.; et al. Stable Isotope Records from Mount Logan, Eclipse Ice Cores and Nearby Jellybean Lake. Water Cycle of the North Pacific Over 2000 337 338 Years and Over Five Vertical Kilometers: Sudden Shifts and Tropical Connections. 339 Geogr. Phys. Quat. 2004, 58 (2-3), 337-352. 340 (20) Holdsworth, G.; Krouse, H. R.; Nosal, M. Ice core climate signals from Mount Logan, 341 Yukon A.D. 1700-1897. In Climate Since AD 1500; Routledge: London and New York, 1992; pp 483–504. 342 (21) Fisher, D.; Osterberg, E.; Dyke, A.; Dahl-Jensen, D.; Demuth, M.; Zdanowicz, C.; 343 344 Bourgeois, J.; Koerner, R. M.; Mayewski, P.; Wake, C.; et al. The Mt Logan Holocene-345 late Wisconsinan isotope record: tropical Pacific-Yukon connections. The Holocene 346 2008, 18 (5), 667–677.

347	(22)	Fisher, D. Connecting the Atlantic-sector and the north Pacific (Mt Logan) ice core stable
348		isotope records during the Holocene. The role of El Niño. <i>The Holocene</i> 2011 <i>21</i> (7)
3/0		
250	(22)	III/. Hammanahmidt C. D. Dawman V. I. Tabatahniak M. D. Lambana C. H. Stanaga
350	(23)	Hammerschmidt, C. K., Bowman, K. L., Tabatchnick, M. D., Lamborg, C. H. Storage
351		bottle material and cleaning for determination of total mercury in seawater. <i>Limnol</i>
352		<i>Oceanogr-Meth</i> 2011 , <i>9</i> , 426–431.
353	(24)	EPA. Method 1631, Revision E: Mercury in Water by Oxidation, Purge and Trap, and
354		Cold Vapor Atomic Fluorescence Spectrometry 2002
355	(25)	NADP National Atmospheric Denosition Program (NRSP-3): NADP Program Office
256	(23)	Illinois State Water Survey 2204 Criffith Dr. Chempsien II 61820 2007
330		minois State water Survey. 2204 Ommun Dr., Champaign, 11 01820, 2007.
357	(26)	Sanei, H.; Outridge, P. M.; Goodarzi, F.; Wang, F.; Armstrong, D.; Warren, K.; Fishback,
358		L. Wet deposition mercury fluxes in the Canadian sub-Arctic and southern Alberta,
359		measured using an automated precipitation collector adapted to cold regions. <i>Atmospheric</i>
360		<i>Environment</i> 2010 , <i>44</i> (13), 1672–1681.
361	(27)	Mao H · Talbot R · Hegarty J · Koermer J Speciated mercury at marine coastal and
362	(= /)	inland sites in New England – Part 2: Relationships with atmospheric physical parameters
262		Atmospheric Chamistry and Dhusics 2012 12 (0) 4191 4206
202	$\langle \mathbf{a} \mathbf{a} \rangle$	Atmospheric Chemistry and Physics 2012 , $12(9)$, $4181-4200$.
364	(28)	Lalonde, J. D.; Amyot, M.; Doyon, MR.; Auclair, JC. Photo-induced Hg(II) reduction
365		in snow from the remote and temperate Experimental Lakes Area (Ontario, Canada). J.
366		Geophys. Res. 2003, 108 (D6), 4200.
367	(29)	Gross, B. H.; Kreutz, K. J.; Osterberg, E. C.; McConnell, J. R.; Handley, M.; Wake, C. P.;
368		Yalcin K Constraining recent lead pollution sources in the North Pacific using ice core
369		stable lead isotones I Geonbus Res-Atmos 2012 117 D16307-D16307
270	(20)	McConnell I. P. Edwards P. Coal hurning laguage toxic heavy metal laguagy in the Arctic
271	(30)	Nicconnent, J. K., Edwards, K. Coar building reaves toxic neavy inetar regacy in the Arctic.
3/1	(2.1)	<i>Proc Natl Acaa Sci USA</i> 2008 , <i>105</i> (34), 12140–12144.
372	(31)	Pilgrim, W.; Hughes, R. Lead, cadmium, arsenic and zinc in the ecosystem surrounding a
373		lead smelter. Environ Monit Assess 1994, 32 (1), 1–20.
374	(32)	Faïn, X.; Ferrari, C. P.; Dommergue, A.; Albert, M. R.; Battle, M.; Severinghaus, J.;
375		Arnaud, L.; Barnola, JM.; Cairns, W.; Barbante, C.; et al. Polar firn air reveals large-
376		scale impact of anthropogenic mercury emissions during the 1970s Proc Natl Acad Sci
377		<i>USA</i> 2009 <i>106</i> (38) 16114_16119
270	(22)	Li C: Cornett I: Willia S: Lam I Margury in Arctia air: The long term trend Sci Tetal
270	(33)	LI, C., Collieu, J., While, S., Lam, J. Mercury in Arctic all. The long-term tiend. Sci Total
3/9		Environ 2009, 407 (8), 2750–2759.
380	(34)	Siebert, L.; Simkin, S. Volcanoes of the world: An illustrated catalogue of Holocene
381		volcanoes and their eruptions. Smithsonian Institution, Global Volcanism Program Digital
382		Information Series, GVP-3. 2002.
383	(35)	Pacyna, J. M.; Pacyna, E. G. An assessment of global and regional emissions of trace
384		metals to the atmosphere from anthropogenic sources worldwide <i>Environ Rev</i> 2001 9 (4)
385		
202	(26)	207. Coolea C. A. Dalaam D. H. Diastar, H. Walfa, A. D. Over three millennia of moreoury.
207	(30)	Cooke, C. A., Balconi, P. H., Biester, H., Wolle, A. P. Over tillee initiennia of mercury
38/	(a =`	pollution in the Peruvian Andes. Proc Natl Acad Sci USA 2009, 106 (22), 8830–8834.
388	(37)	Beal, S. A.; Kelly, M. A.; Stroup, J. S.; Jackson, B. P.; Lowell, T. V.; Tapia, P. M. Natural
389		and anthropogenic variations in atmospheric mercury deposition during the Holocene near
390		Quelccaya Ice Cap, Peru. Global Biogeochem Cy 2014, 28, 437–450.
391	(38)	Vandal, G. M.; Fitzgerald, W. F.; Boutron, C. F.; Candelone, JP. Variations in mercurv
392	(-)	deposition to Antarctica over the past 34 000 years <i>Nature</i> 1993 362 (6421) 621–623
		$= r^{$
		16

- 393 (39) Amos, H. M.; Jacob, D. J.; Streets, D. G.; Sunderland, E. M. Legacy impacts of all-time 394 anthropogenic emissions on the global mercury cycle. Global Biogeochem Cy 2013, 27 395 (2), 410-421.
- 396 (40) Harris, R. C.; Rudd, J. W. M.; Amyot, M.; Babiarz, C. L.; Beaty, K. G.; Blanchfield, P. J.; 397 Bodaly, R. A.; Branfireun, B. A.; Gilmour, C. C.; Gravdon, J. A.; et al. Whole-ecosystem 398 study shows rapid fish-mercury response to changes in mercury deposition. Proc Natl 399 Acad Sci USA 2007, 104 (42), 16586–16591.
- 400 (41) Beal, S. A.; Jackson, B. P.; Kelly, M. A.; Stroup, J. S.; Landis, J. D. Effects of historical 401 and modern mining on mercury deposition in southeastern Peru. Environ. Sci. Technol. 402 **2013**, *47* (22), 12715–12720.
- (42) Slemr, F.; Brunke, E.-G.; Ebinghaus, R.; Kuss, J. Worldwide trend of atmospheric 403 404 mercury since 1995. Atmospheric Chemistry and Physics 2011, 11 (10), 4779-4787.
- Cole, A. S.; Steffen, A.; Pfaffhuber, K. A.; Berg, T.; Pilote, M.; Poissant, L.; Tordon, R.; 405 (43) 406 Hung, H. Ten-year trends of atmospheric mercury in the high Arctic compared to 407 Canadian sub-Arctic and mid-latitude sites. Atmos. Chem. Phys. 2013, 13 (3), 1535-1545.
 - Clarke, G. K. C.; Jarosch, A. H.; Anslow, F. S.; Radic, V.; Menounos, B. Projected (44)
- 408 409 deglaciation of western Canada in the twenty-first century. *Nature Geosci* 2015, 8 (5), 410 372-377.
- 411 (45) Lenaerts, J. T. M.; van Angelen, J. H.; van den Broeke, M. R.; Gardner, A. S.; Wouters, 412 B.; van Meijgaard, E. Irreversible mass loss of Canadian Arctic Archipelago glaciers. 413 Geophys. Res. Lett. 2013, 40 (5), 870-874.
- 414 (46) Kalnay, E. The NCEP/NCAR Reanalysis 40-year Project. B Am Meteorol Soc 1996, 77, 437-471. 415
- 416 Tables
- 417 **Table 1.** Descriptive statistics for detectable Hg_T concentrations in samples, procedural blanks, and outer
- 418 core material removed during decontamination of Mount Logan firn and ice. Concentrations for samples
- 419 and the outer core are procedural blank corrected, whereas procedural blank concentrations are
- 420 instrumental blank (0.02 pg g^{-1}) corrected.

	$Hg_T (pg g^{-1})$						
	Firn			Ice			
	Samples	Proc. Blank		Samples	Proc. Blank	Outer Core	
Max	3.52	0.33		1.06	0.34	9.41	
Min	0.13	0.16		0.02	0.01	0.35	
Mean	0.91	0.24		0.21	0.09	5.63	
Median	0.66	0.23		0.14	0.04	5.41	
Std. dev	0.82	0.06		0.23	0.10	2.29	
n	37	12		51	43	13	

Figures 422



423

424 Figure 1. Location of Mount Logan (red triangle) in the St. Elias Mountains on the border of Alaska and

Yukon, Canada. Arrows represent annual average vector wind at 500 mb from AD 1948–1998.⁴⁶ 425



426

427 Figure 2. 20th Century trends in atmospheric Hg deposition at Mount Logan compared with ice core

428 records of Pb pollution and the longest available measurements of remote atmospheric Hg concentrations. 429 a) Mount Logan Hg_T fluxes (blue points) with 1σ error bars and LOESS smoother (red line). Grey

430 shading denotes two periods of elevated Hg_T fluxes. b) Annually averaged Pb concentrations in the Mount

431

Logan ice core¹⁸ and the Eclipse Icefield ice core,²⁹ both in the St. Elias Mountains, and in the Act2 ice core in southern Greenland.³⁰ c) Modeled firn-air Hg⁰ concentrations in a core from Summit in central 432

Greenland modified from Faïn et al.³² d) Total filterable (particulate) Hg concentrations in air samples
 from Resolute in Arctic Canada modified from Li et al.³³



435

436 **Figure 3.** Multi-century Hg_T records from ice cores compared with estimates of primary anthropogenic 437 emissions used in recent global Hg models. a) Mount Logan Hg_T fluxes (blue points) with 1 σ error bars,

438 LOESS smoother (red line), and inset with adjusted y-axis for the Preindustrial Period. b) Hg_T fluxes in

the Upper Fremont Glacier ice core calculated using an assumed constant accumulation rate of 800 kg m⁻² a⁻¹ modified from Schuster et al.¹⁴ c) Estimated primary anthropogenic Hg emissions from industrial and

440 a modified from Scrutster et al.,⁷ and additional emissions from commercial Hg use modified

442 from Horowitz et al.⁸



82x29mm (300 x 300 DPI)